organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Veronica Åberg,^a Dan Boström,^b Andreas Fischer^c and Fredrik Almqvist^a*

^aOrganic Chemistry, Department of Chemistry, Umeå University, 901 87 Umeå, Sweden, ^bInorganic Chemistry, Department of Chemistry, Umeå University, 901 87 Umeå, Sweden, and ^cInorganic Chemistry, Royal Institute of Technology, 100 44 Stockholm, Sweden

Correspondence e-mail: fredrik.almqvist@chem.umu.se

Key indicators

Single-crystal X-ray study T = 120 KMean σ (C–C) = 0.006 Å R factor = 0.044 wR factor = 0.103 Data-to-parameter ratio = 14.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Synthesis and absolute configuration of methyl (-)-(3*R*)-8-(4-bromophenyl)-7-(naphthalen-1-yl-methyl)-5-oxo-2,3-dihydro-5*H*-thiazolo[3,2-a]pyridine-3-carboxylate

The title molecule, $C_{26}H_{20}BrNO_3S$, contains a ring-fused 2-pyridinone framework substituted with a 4-bromo-phenyl-, a naphthalen-1-ylmethyl and a methoxycarbonyl substituent. The main goal of this work was to confirm the stereochemistry for the methoxycarbonyl substituent, which proved to be 3R. Moreover, the 4-bromophenyl substituent was shown to be rotated out of the plane of the 2-pyridinone ring, with a torsion angle of $61.2 (5)^{\circ}$. To allow the best packing arrangement, the naphthalen-1-ylmethyl substituent is positioned to mediate an intermolecular π - π interaction.

Received 19 June 2002 Accepted 28 June 2002 Online 12 July 2002

Comment

Heterocycles having a 2-pyridinone framework are extensively studied classes of compounds, due partly to their diverse biological activities. These range from antibacterial (Casinovi *et al.*, 1968; Dolle & Nicolaou, 1985; Rigby & Balasubramanian, 1989) and antifungal (Cox & Hagan, 1991) agents to free-radical scavengers (Teshima *et al.*, 1991; Rice-Evans & Burdon, 1994). Ring-fused 2-pyridinones have also attracted attention as angiotensin-converting enzyme (ACE) inhibitors (Mynderse *et al.*, 1985; Hunt *et al.*, 1988; O'Connor & Somers, 1985), as well as inhibitors of A β -peptide aggregation (Kuner *et al.*, 2000; Thorsett & Latimer, 2000), which is believed to play an important role in amyloid formation in Alzheimer's disease.



As a result of its anticancer properties, the natural product (S)-camptothecin, isolated from *Campthoteca acuminata* by Wall *et al.* (1966), is one of the most studied ring-fused 2-pyridinones (Comins & Nolan, 2001). The structure of (S)-camptothecin was elucidated two years later by X-ray crystallography (McPhail & Sim, 1968). Previously, we reported a new ketene–imine cycloaddition method for the synthesis of

 \odot 2002 International Union of Crystallography Printed in Great Britain – all rights reserved



Figure 1 A view of the molecular structure of (5), showing 50% probability displacement ellipsoids.

substituted optically active ring-fused 2-pyridinones (Emtenäs *et al.*, 2001). Although having high enantioselectivity in the reaction, the stereochemistry remained unconfirmed. Therefore, from previous experience of bromine-containing aromatics being easier to crystallize, we synthesized the title compound, (5). Additionally, the Br atom facilitates the determination of the absolute configuration, due to the high anomalous scattering of Mo radiation.

The crystal structure determination of (5) confirmed the 3R configuration. In addition, the 8-(4-bromophenyl) substituent was shown to be rotated out of the plane, with a torsion angle of 61.2 (5)° (atoms C13, C8, C6 and C7; see Fig. 1). Although somewhat more pronounced, this arrangement has previously been observed by Jones *et al.* (2000), where their corresponding phenyl substituent was almost perpendicular to the plane, with a torsion angle of 103° . There is a possible weak hydrogen-bond interaction between O1 and C1 in a neighbouring molecule, 3.085 (6) Å. Otherwise, there are no intermolecular interactions shorter than van der Waals distances, except for one, C19–H(C10) of 2.75 and 0.15 Å less than the van der Waals distance. As can be seen in Fig. 2, the naphthalen-1-ylmethyl substituent is positioned to mediate an intermolecular π - π interaction.

Experimental

For the preparation of methyl 4-(R)-2-(4-bromobenzyl)-4,5-dihydrothiazole-4-carboxylate, (3), dry HCl(g) was passed through a stirred solution of (1) (5.12 g, 26 mmol) in EtOH-CH₂Cl₂ (1:1, 20 ml) for 4 h at 273 K. The solution was allowed to reach room



Figure 2 A packing diagram of (5), viewed down the *b* axis.

temperature overnight and was then concentrated, giving (2) as colourless crystals (7.27 g, quant.), which were used in the next step without further purification. Et₃N (2.0 ml, 14 mmol) was added dropwise to a stirred suspension of L-cysteine methyl ester hydrochloride (3.02 g, 11 mmol) in CH₂Cl₂ (15 ml) at 273 K. After 30 min, (2) was added and the suspension was allowed to reach room temperature. After stirring overnight, the suspension was diluted with CH₂Cl₂ (25 ml) and washed with water, aqueous saturated NaHCO₃ and brine. The organic layer was dried (Na₂SO₄), filtered and concentrated. Flash column chromatography (heptane–EtOAc, 1:1) gave the thiazoline (3) as an oil (3.37 g, quant.). $[\alpha]_D$ +53° (*c* 0.91, CHCl₃).

For the synthesis of the title compound, (5), dry HCl (g) was passed through a 273 K solution of (3) (2.08 g, 7 mmol) and (4) (Emtenäs *et al.*, 2001; 2.71 g, 9 mmol) in 1,2-dichloroethane (45 ml) for 15 min. The solution was stirred for 12 h at 337 K before an additional amount of (4) (1.30 g, 4 mmol) was added. After 3 h, the solution was allowed to reach room temperature and was then diluted with CH₂Cl₂ and washed with water, aqueous saturated NaHCO₃ and brine. The organic phase was dried (Na₂SO₄), filtered and concentrated, Flash column chromatography (heptane–EtOAc, 1:4) gave (5) as a white solid (1.79 g, 53%). The product (5) was crystallized from EtOH [m.p. 475–476 K; $[\alpha]_D$ –134° (*c* 1.26, CHCl₃)].

Crystal data	
$C_{26}H_{20}BrNO_{3}S$ $M_{r} = 506.40$ Monoclinic, P2 ₁ a = 9.5502 (5) Å b = 8.6443 (5) Å c = 13.1213 (6) Å $\beta = 91.028$ (3)° V = 1083.05 (10) Å ³ Z = 2	$D_x = 1.553 \text{ Mg m}^{-3}$ Mo K\$\alpha\$ radiation Cell parameters from 8301 reflections $\theta = 4.5-27.5^{\circ}$ $\mu = 2.02 \text{ mm}^{-1}$ T = 120 K Plate, colourless $0.25 \times 0.13 \times 0.03 \text{ mm}$
Data collection	
KappaCCD diffractometer ω scans Absorption correction: numerical (Herrendorf & Bärnighausen, 1997) $T_{min} = 0.566, T_{max} = 0.815$ 8301 measured reflections	4234 independent reflections 3754 reflections with $l > 2\sigma(l)$ $R_{int} = 0.052$ $\theta_{max} = 27.5^{\circ}$ $h = -12 \rightarrow 12$ $k = -11 \rightarrow 11$ $l = -16 \rightarrow 17$

Refinement Refinement on F^2 $w = 1/[\sigma^2(F_o^2) + (0.0225P)^2]$ $R[F^2 > 2\sigma(F^2)] = 0.044$ wR(F²) = 0.103 + 1.1354P] where $P = (F_0^2 + 2F_c^2)/3$ S=1.04 $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 0.76 \text{ e } \text{\AA}^{-3}$ 4234 reflections $\Delta \rho_{\rm min} = -0.52 \text{ e } \text{\AA}^{-3}$ 289 parameters H-atom parameters constrained Absolute structure: (Flack, 1983), 1582 Friedel pairs Flack parameter = 0.013 (10)

The H atoms were refined using a riding model.

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *HKL SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO* (Otwinowski & Minor, 1997) and *SCALEPACK*; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ATOMS* (Dowty, 1998); software used to prepare material for publication: *SHELXL*97.

This work was funded by grants from the Knut and Alice Wallenberg Foundation, the Foundation for technology transfer in Umeå and the Swedish Research Council.

References

- Casinovi, C. G., Grandolini, G., Mercantini, R., Oddo, N., Olivieri, R. & Tonolo, A. (1968). *Tetrahedron Lett.* pp. 3175–3178.
- Comins, D. L. & Nolan, J. M. (2001). Org. Lett. 3, 4255-4257.

Cox, R. J. & O'Hagan, D. (1991). J. Chem. Soc. Perkin Trans. 1, pp. 2537–2540.

Dolle, R. E. & Nicolaou, K. C. (1985). J. Am. Chem. Soc. 107, 1691–1694. Dowty, E. (1998). ATOMS. Shape Software, 521 Hidden Valley Road,

Kingsport, Tennessee, USA.

Emtenäs, H., Alderin, L. & Almqvist, F. (2001). J. Org. Chem. 66, 6756–6761. Flack, H. D. (1983). Acta Cryst. A39, 876–881.

Herrendorf, W. & Bärnighausen, H. (1997). *HABITUS*. Giessen University, Germany.

- Hunt, A. H., Mynderse, J. S., Samlaska, S. K., Fukuda, D. S., Maciak, G. M. Kirst, H. A., Occolowitz, J. L., Swartzendruber, J. K. & Jones, N. D. (1988). J. Antibiot. 41, 771–779.
- Jones, R. C., Dimopoulos, P., Cloes, S. C., Light, M. E. & Hursthouse, M. B. (2000). J. Chem. Soc. Perkin Trans. 1, pp. 2331–2342.
- Kuner, P., Bohrmann, B., Tjernberg, L. O., Näslund, J., Huber, G., Celenk, S., Grüninger-Leitch, F., Richards, J. G., Jakob-Roetne, R., Kemp, J. A. & Nordstedt, C. (2000). J. Biol. Chem. 275, 1673–1678.
- McPhail, A. T. & Sim, G. A. (1968). J. Chem. Soc. B, pp. 923–928.
- Mynderse, J. S., Samlaska, S. K., Fukuda, D. S., Du Bus, R. H. & Baker, P. J. (1985). J. Antibiot. **38**, 1003–1007.
- Nonius (1998). COLLECT. Nonius BV, Delft, The Netherlands.
- O'Connor, S. & Somers, P. (1985). J. Antibiot. 38, 993–996.
- Otwinowski, Z. & Minor, W. (1997). Methods in Enzymology Vol. 276, Macromolecular Crystallography, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Rice-Evans, C. A. & Burdon, R. H. (1994). Free Radical Damage and its Control. Amsterdam: Elsevier.
- Rigby, J. & Balasubramanian, N. (1989). J. Org. Chem. 54, 224-228.
- Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.
- Teshima, Y., Shin-ya, K., Shimazu, A., Furihata, K., Chul, H. S., Furihata, K., Hayakawa, Y., Nagai, K. & Seto, H. (1991). J. Antibiot. 44, 685–687.
- Thorsett, E. D. & Latimer, L. H. (2000). Curr. Opin. Chem. Biol. 4, 377-382.
- Wall, M. E., Wani, M. C., Cook, C. E., Palmer, K. H., McPhail, A. T. & Sim, G. A. (1966). J. Am. Chem. Soc. 88, 2888–2890.